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2017 CHEMISTRY EXAM TSSM 2017 Page 3 of 13 Question 4 Answer: C Explanation: 3KOH (aq) + AlCl 3 (aq) 3KCl (aq) + Al(OH) 3 (s) 3Zn(NO 3) 2 (aq) + 2K 3 PO 4 (aq) Zn 3 (PO 4) 2 (s) + 6KNO 3 (aq) Sr(OH) 2 + 2NH 4 CH 3 COO (aq) 2NH 4 OH + Sr(CH 3 COO) 2 (aq) Na 2 SO 4 (aq) + Pb(NO 3) 2 (aq) PbSO 4 (s) + 2NaNO 3 (aq) Question 5

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Explanation: Use kinematics equations to find the initial vertical velocity, given vvert= 0 at 1.8 sec after launch: v =u+at 0 =u−10×1.8 u =18ms−1. Then, use trigonometry to determine v at 35o: v = 18 cos(55) =31.4ms−1. Question 18. Answer: R = 46 m Explanation: vhoriz=v xcos(35) =25.7ms−1. x =vh x t = 46.26 m.

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Units 3 & 4 – Written examination (TSSM' s 2015 trial exam updated for the current study design) SOLUTIONS . PHYS EXAM TSSM 2017 Page 2 of 17 Section A - Multiple Choice (1 mark each) Question 1 Answer: C Explanation: The density of lines at a specific location in space reveals information about the strength of ...

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2011 Trial Examination SOLUTIONS SECTION A – Core Area of Study 1 – Motion ... 2011 PHYS EXAM 2 TSSM 2011 Page 5 of 21 Question 12 Answer: 6.7 x 104 N m-1 Explanation: The gravitational potential energy at 5 m ( mgh = 0.15 x 10 x 5 = 7.5 J) is all converted to kinetic ...

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2010 Trial Examination SOLUTIONS SECTION A – Core Area of Study 1- Electric Power Question 1 Answer: B ... 2010 PHYS EXAM 2 TSSM 2010 Page 3 of 16 Question 10 Answer: 300 turns Explanation: Current at the generator is 10000/250 = 40 A. Current in the lines is 10 A. Thus the transformer is a step-up in the ratio of 1:4 ...

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TSSM 2018 Page 1 of 1 2018 Trial Examination Letter STUDENT NUMBER THIS BOX IS FOR ILLUSTRATIVE PURPOSES ONLY CHEMISTRY Unit 1 – Written examination Reading time: 15 minutes Writing time: 1 hour and 30 minutes QUESTION & ANSWER BOOK Structure of book Section Number of questions Number of questions to be answered Number of marks A 20 20 20 B 6 6 66 Total 86 Students are ...

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2017 CHEMISTRY EXAM TSSM 2017 Page 2 of 13 SECTION A: Multiple-choice questions (1 mark each) Question 1 Answer: D Explanation: Ice is less dense than water as the regular array of water molecules formed as a result of hydrogen bonding causes gaps to form. Question 2 Answer: C Explanation: 2140 J = 2.14 kJ n (water) = 2.14/6.0 = 0.357 mol m (water) = 0.357 x 18.0 = 6.42 g Question 3 Answer: D ...

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TSSM 2011 Page 1 of 19 PHYSICS Unit 1 – Written examination 1 2011 Trial Examination SOLUTIONS SECTION A Area of Study 1 – Nuclear Physics and Radioactivity Question 1 Answer:A Explanation: As per definition of each type of radioactive decay. Alpha is the heaviest and doubly

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