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TOPIC 9.1 (Part 1) -

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Solution: A solution is a homogeneous mixture of two (or more) substances, the composition of which may vary between certain limits. A solution consisting of two components is called binary solution. The component which is present in large quantity is called

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solvent and the component which is small in quantity is called solute. If both components are in same physical state.

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A solution is a homogeneous mixture of two or more components in which the particle size is

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smaller than 1 nm.

Common examples of solutions are the sugar in water and salt in water solutions, soda water, etc. In a solution, all the components appear as a single phase.

There is particle homogeneity i.e. particles are evenly distributed.

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Solution - Definition,
Properties, Types,
Videos & Examples
Chemistry 108

Lecture Notes

Chapter 7: Solutions 3

Solutions □ The primary ingredient in a solution is called the _____. □ The other ingredients are the _____ and are said to be dissolved in the solvent. □ Water is the

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most common solvent. □ Water is a unique solvent because so many substances can dissolve in it.

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definitions □ Solution: homogenous mixture
□ Solute: substance that is dissolved □
Solvent:... 3. Types of Solutions 4. □
Saturated solutions are at dynamic equilibrium □ the solute is dissolving and crystallizing at an equal ...

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Solutions Solution is a
homogeneous mixture
of two or more
substances in same
or different physical
phases. The
substances forming
the solution are called
components of the

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Chemistry -

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Solution is a homogeneous mixture of two or more substances in same or different physical phases. The substances forming the solution are called components of the solution. On the basis of number of components a

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A solution of two components is called binary solution. Solute and Solvent In a binary solution, solvent is the component which is present in large quantity while the other

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1. A solution is a homogeneous mixture of two or 9. more chemically non-reacting substances. The components of a solution generally cannot be separated by filtration, settling or centrifuging. 2. A solution may be classified as solid, liquid or a gaseous solution. 3.

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that contain

questions, exercise

and key points of the

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The minor component of the solution is called the solute. In the present example, sugar is the solute.

The major component of the solution is called the solvent. In this case water is the solvent. Solutions can also be formed by mixing together many

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different phases of matter. For instance, air is a solution.

Composition of
Solutions: Solution
Composition |
SparkNotes
A chemical solution
exhibits several
properties: A solution
consists of a
homogeneous
mixture. A solution is

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composed of one phase (e.g., solid, liquid, gas). Particles in a solution are not visible to the naked eye.

Solution Definition in
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Notes Chapter 2. A
solution comprises a
solute and a solvent.

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It is defined as a homogeneous mixture of two or more than two substances. They can be classified into three types: Solid solutions. Gaseous solutions. Liquid solutions. Molarity, mole fraction, percentages, and molality are the terms used to express the concentration of a

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Chapter 2 Solutions
Chemistry. From
aluminum to xenon,
we explain the
properties and
composition of the
substances that make
up all matter.

Chemistry Study
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What makes up a solution?

- A solution has two components: the solute and the solvent.

- The solvent is the substance in greater amount (it does the dissolving).

- The

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In Chemistry
solute is considered
to be the dissolved
substance.

Chemistry
30S Unit 4 - Solutions

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Solutions which
contain two
components in it are
called Binary
Solutions. Substances
which are used to
prepare a solution are

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called as Components. The component that is present in the largest quantity is known as Solvent. Solvent determines the physical state in which solution exists.

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substances forming the solution are called components of the solution. On the basis of number of components a solution of two components is called binary solution.

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