

## Pogil Equilibrium Answer Key

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Introduction to POGIL **Guided Inquiry Labs For AP Chemistry** Pogil Equilibrium Answer Key

The following equation describes a system that is at equilibrium: 2H<sub>2</sub> (g) + O<sub>2</sub> (g) ⇌ 2H<sub>2</sub>O (g) + energy (heat) In Table 1 apply Le Chatelier's Principle and indicate the direction of the shift in equilibrium if the indicated stress is applied to the reaction system. (The first one is completed for you.) Key Questions 1. Complete the following table:

Mrs. Zuberbuehler - Mrs. Zuberbuehler

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Answer Key -Watch: Reaction Quotient Tuesday 10/17 -POGIL -Notes: Reaction Quotient Powerpoint (Slides 15-19) -Watch: LeChatelier's Principle Equilibrium Disturbance -Read: Ch 13.3 -Problem Set: Rxn Quotient #18 (NOT 17, the #s shifted!) Answer Key Wednesday 10/18

Unit 3: Equilibrium - Ms. Burlison

chi-square analysis to determine if the population is in Hardy-Weinberg equilibrium. In the original population: P = 80/120 = 0.66 q 40/120 = 0.33 50 The predicted genotype frequencies for the population once it has reached Hardy- Weinberg equilibrium are: p<sup>2</sup> 0.4356 2pq = 0.4356 0.1089

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