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4.2.2 How much energy is released when bonds form to produce 2 mol of H 2 O and 1 mol of CO 2? (4) 4.2.3 Hence calculate the energy transferred (ΔH) for the combustion of 1 mol CH 4. (2) 4.2.4 Draw a labelled energy profile for this reaction. Include the labels: ENTHALPY CHANGE (ΔH) ACTIVATION ENERGY (E a) (6)

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