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11) Which molecule has a dipole moment of zero? A)  $\text{CHCl}_3$  B)  $\text{CH}_2\text{Cl}_2$  C)  $\text{ClHC}=\text{CH}_2$  D) *trans*- $\text{ClHC}=\text{CHCl}$  E) None of these choices. Answer: D Topic: Molecular geometry, polarity Section Reference 1: 1.5 and 2.3 Difficulty: Medium 12) Which molecule would have a dipole moment greater than zero? A)  $\text{BeCl}_2$  B)  $\text{BCl}_3$  C)  $\text{CO}_2$  D)  $\text{H}_2\text{O}$  E)  $\text{CCl}_4$  Answer: D

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