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CHAPTER 3 REVIEW Atoms: The  
Building Blocks of Matter SECTION 3  
SHORT ANSWER Answer the following

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Answers in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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3 Atoms: The Building Blocks of Matter

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CHAPTER 9 REVIEW Stoichiometry  
MIXED REVIEW SHORT ANSWER

Answer the following questions in the

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Answers Chapter 14  
space provided. 1. Given the following  
equation:  $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$   
a. What is the value of the coefficient x in this equation? 4  
b. What is the molar mass of  $C_3H_4$ ? 40.07 g/mol  
c. What is the mole ratio ...  
mol  $O_2$ :1 mol  $H_2O$

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Answers Chapter 4  
Terms in this set (30) chemical bond. the attractive force that holds atoms or ions together. ionic bonding. a force that attracts electrons from one atom to another, which transforms a neutral atom into an ion.

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