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periodic functions of their atomic numbers.  
periodic table. an arrangement of the  
elements in order of their atomic numbers  
so that elements with similar properties fall  
in the same column, or group. lanthanides.

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CHAPTER 5 REVIEW The Periodic Law

MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided. 1. Consider the neutral atom with

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53 protons and 74 neutrons to answer the following questions. 53 a. What is its atomic number? 127 b. What is its mass number? atomic number c. Is the element ' s position in a modern periodic table determined by

## 5 The Periodic Law

Modern Chemistry Chapter 5. STUDY.

PLAY. periodic law. The physical and chemical properties of the elements are periodic functions of their atomic numbers. lanthanide. The 14 elements with the atomic numbers 58-71. actinide. The 14 elements with the atomic numbers 90-103. alkali metals.

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Modern Chemistry 33 Chapter Test Name  
Class Date Chapter Test B, continued 15.  
The energy state of an atom is called its ground state. 16. The number of waves that

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pass a point in one second is called. 17.  
When an electron drops from a higher-energy state to a lower-energy state, a(n) spectrum is produced. 18.

Assessment Chapter Test B

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PLAY. Actinide. Any of the elements of the actinide series which have atomic numbers from 89 through 103. Alkali metals. One of the elements of Group 1 of the periodic table. Alkaline earth metals. One of the elements of group 2 of the periodic table. Anion.

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