

## Modeling Chemistry U6 Ws 5 V1 0 Agunot

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IB Chemistry Topic 5 Energetics 5.1 Q=mcΔT - heat and calculations

Balancing Chemical Equations Practice Problems *Investigating the Periodic Table with Experiments - with Peter Wothers Representing Ions and Formula Units* WS#3 *DNA Replication (Updated)*

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The Periodic Table: Crash Course Chemistry #4 How Small Is An Atom? Spoiler: Very Small. *The Origin of the Elements*

Kinetics: Initial Rates and Integrated Rate Laws

IB Chemistry Topic 5 Energetics 5.2 Hess's Law with enthalpy of formation and enthalpy of combustion *DNA vs RNA (Updated)* *AP Chem CH7 Atomic Structure and Periodicity* **Heat of Combustion of Magnesium Hess' Law**

**Lab Hess' Law Cycles Involving Enthalpies of Combustion** 5.1 Standard enthalpy change of combustion (SL) *Protein Synthesis (Updated)* *Redox Reactions: Crash Course Chemistry #10 Biomolecules (Updated)* *Models of the Atom Timeline* *Zumdahl Chemistry 7th ed. Chapter 7 (Pt. 1)* *Amazon Empire: The Rise and Reign of Jeff Bezos (full film)* | *FRONTLINE Separation of Substances | Class 6 Science Sprint for Final Exams | Chapter 5 @Vedantu Young Wonders Workshop on Development: Lock-ins, Traps and Catch Up - Parte 4* **Chem 315 Fall 2020 Lecture 14 for Sept 9 Modeling Chemistry U6 Ws 5**

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Modeling Chemistry 2 U6 ws 3 v1.0 3. Next to the formula, give the name of the following simple binary ionic compounds. a. Na<sub>2</sub>O b. MgCl<sub>2</sub> c. CaBr<sub>2</sub> d. Al<sub>2</sub>S<sub>3</sub> e. CsBr f. AgF g. Na<sub>3</sub>N h. K<sub>2</sub>O 5. Write the name of these ionic substances next to the formula, using a Roman numeral to specify the charge of the cation.

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