

## Chapter 12 Basics Of Chemistry

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chemistry. **0** Describe oxidation and reduction (redox) reactions. **0** Discuss the different forms of matter: elements, compounds, and mixtures. **0** Explain the difference between solutions, suspensions, and emulsions. **0** Explain pH and the pH scale.

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organic chemistry. The study of substances that contain the element carbon. oxidation. A chemical reaction that combines a substance with oxygen to produce an oxide. oxidation-reduction. Also known as redox; a chemical reaction in which the oxidizing agent is reduced and the reducing agent is oxidized. oxidizing agent.

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A chemical reaction that combines a substance with oxygen to produce an oxide. oxidation-reduction. (aka redox) a chemical reaction in which the oxidizing agent is reduced (by losing oxygen) and the reducing agent is oxidized (by gaining oxygen). oxidizing agent. Substance that releases oxygen.

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A stable physical mixture of 2 or more substances. ALKANOLAMINES. Alkaline substances used to neutralize acids or raise the PH of many hair products. EMULSION. An unstable physical mixture of 2 or more immiscible substances, plus a special ingredient called a "emulsifier".

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