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~~3.7-3.10 Solutions,
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~~Intermolecular Forces,~~

~~Solids, Liquids, and Gases~~

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AP Chemistry Free-Response
Questions (FRQ) – Past
Prompts. Published on Apr
19, 2020. [view the 2020 ap](#)

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chem study plan. We've compiled a sortable list of all AP Chemistry past prompts! The AP Chem FRQs are 50% of the exam and include 3 long questions, which are worth 10 points, and 4 short questions, each

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worth 4 points.

AP Chemistry Free-Response Questions (FRQ) - Past Prompts ...

A student obtains a solution that contains an unknown concentration of Fe. 2+

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(aq). To determine the concentration of Fe^{2+} (aq) 2018 AP [®] CHEMISTRY FREE-RESPONSE QUESTIONS. in the solution, the student titrates a sample of the solution with MnO_4^- (aq), which converts Fe^{2+}

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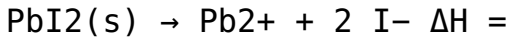
(aq) to. Fe. 3+ (aq), as
represented by the following
equation. 5 Fe. 2+ (aq) +
MnO . 4 -

AP Chemistry 2018 Free- Response Questions

AP* Solution Chemistry Free

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Response Questions page 6
1995 Lead iodide is a dense, golden yellow, slightly soluble solid. At 25°C, lead iodide dissolves in water forming a system represented by the following equation.



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+46.5 kilojoules

AP* Solution Chemistry Free Response Questions

May 16, 2017 / in AP

Chemistry / by erinelson.

The free-response section of
the AP Chemistry test can be

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daunting. It consists of seven questions—three long and four short—that you will have 105 minutes to answer. It is suggested that you spend 7 minutes on each short questions and 25 minutes on each long

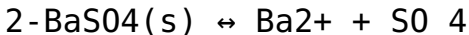
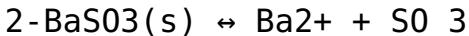
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...

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Response KEY page 3 1980 (a)
 $\text{BaCO}_3(\text{s}) \leftrightarrow \text{Ba}^{2+} + \text{CO}_3^{2-}$

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2-Dissolving takes place if equilibrium is shifted to the right.

$$\text{CO}_3^{2-} + \text{H}^+ \rightarrow \text{HCO}_3^-$$
$$\text{HCO}_3^- + \text{H}^+ \rightarrow \text{H}_2\text{CO}_3 + \text{CO}_2(\text{g})$$
$$\text{SO}_3^{2-} + \text{H}^+ \rightarrow \text{HSO}_3^-$$
$$\text{HSO}_3^- + \text{H}^+ \rightarrow \text{H}_2\text{SO}_3 + \text{SO}_2(\text{g})$$

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**AP Chemistry: Solutions and
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in college-level learning.

Not all free-response questions on this page reflect the current exam, but the question types and the topics are similar,

AP Chemistry Past Exam

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**AP Chemistry: Solids,
Liquids, and Intermolecular
Forces ...**

temperature reached by the
solution is 35.6 C. (i)

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Calculate the magnitude of the heat absorbed by the solution during the dissolution process, assuming that the specific heat capacity of the solution is $4.18 \text{ J}/(\text{g}\cdot^{\circ}\text{C})$. Include units with your

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$$\text{answer. } q = mc \Delta T = (110.0 \text{ g})(4.18 \text{ J/(g} \cdot \text{ }^\circ\text{C)})(35.6 \text{ }^\circ\text{C} - 15.0 \text{ }^\circ\text{C}) = 9,470 \text{ J} = 9.47 \text{ kJ}$$

AP Chemistry Scoring Guidelines, 2016 - College Board

Properties of Solutions 11

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00 total A B A= + = + χ χ^P P P
P P PA B B (same as Dalton's Law) Exercise 7 Calculating the VP of a Solution Containing Two Liquids. A solution is prepared by mixing 5.81 g acetone (C_3H_6O , molar mass = 58.1

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g/mol) and 11.9 g chloroform (HCCl₃, molar mass = 119.4 g/mol).

AP* Chemistry PROPERTIES OF SOLUTIONS

A student is given 50.0 mL of a solution of Na₂CO₃ of

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unknown concentration. To determine the concentration of the solution, the student mixes the solution with excess 1.0 M $\text{Ca}(\text{NO}_3)_2$ (aq), causing...

2019 AP Chemistry Free

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Response #3 Solutions - YouTube

2020 Free-Response Questions. By Thanksgiving, 2020 free-response questions will be in the AP Classroom question bank for teachers to assign to students as

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homework or in class, and will not require secure assessment. These questions will be updated, where possible, to best match the format of free-response questions in the course and exam description and on

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AP Chemistry Exam - AP Central | College Board

You should spend a maximum of 20 minutes on each long free-response question and 10 minutes on each short

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free-response question. Keep an eye on the time so it doesn't get away from you! Here are some basic steps you should take to solve free-response questions:

AP Chemistry FRQ: How to Ace

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the Free Response Questions

AP* Stoichiometry Free

Response Questions page 4

(b) In a reaction vessel,
0.600 mol of $\text{Ba}(\text{NO}_3)_2$ (s)
and 0.300 mol of H_3PO_4
(aq) are combined with
deionized water to a final

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volume of 2.00 L. The reaction represented below occurs.

$$3 \text{Ba}(\text{NO}_3)_2 (\text{aq}) + 2 \text{H}_3\text{PO}_4 (\text{aq}) \rightarrow \text{Ba}_3(\text{PO}_4)_2 (\text{s}) + 6 \text{HNO}_3 (\text{aq})$$

Calculate the mass of $\text{Ba}_3(\text{PO}_4)_2 (\text{s})$ formed.

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AP* Stoichiometry Free Response Questions

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response 3a. ... Is commonly added to diet soft drinks as a preservative. A stock solution of potassium sorbate, dissolved in an aqueous solution here, of known concentration must be prepared. A student titrates

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45 millileters of the stock solution with one point two five molar hydrochloric acid using both an ...

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2015 AP Chemistry free response 5a: Finding order of reaction . 2015 AP Chemistry free response 6. 2015 AP Chemistry free response 7. Video transcript - [Voiceover] Ethene, C_2H_4 , molar mass of 28.1 grams per

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mole, may be prepared by the dehydration of ethanol, C_2H_5OH , molar mass 46.1 grams per mole, using a solid catalyst. A setup for the lab ...

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**response 2a (part 1 of 2)
(video ...**

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